

AMENDMENTS TO THE SPECIFICATION:

Please amend the specification as follows:

Please replace the paragraph [0015] on Page 4 with the following:

[0015] Since a stable titanium-containing aqueous solution was obtained in the presence of the amines, etc. and the ketones, etc., it was at first, presumed that a remarkable effect of a mixed system of the amine, etc. and the ketone, etc. was attributable to the formation of an imine product having a structure like diethanolamine (DEA) ~~functioning~~ functioning as a multidentate ligand-chelating ligand. However, ammonia or N, N-dimethylethanolamine (DMEA) cannot form such a compound, but ammonia and DMEA are ~~compounds~~ compounds highly effective in obtaining the stable titanium-containing aqueous solution. Particularly, the latter is a N, N-dimethyl derivative of monoethanolamine (MEA), so that it does not form an imine with a ketone. This result strongly suggests that the amine used exhibited some unexpected effect.

Please replace the paragraph [0041] on Page 13 with the following:

[0041] The aqueous solutions obtained according to the producing method of the present invention have another ~~advantageous~~ advantage that the ~~solution~~ solutions are useful for the preparation of titanium-mixed oxide materials containing elements which salts are soluble in water.